

## 4,5-Bis(phenylsulfonylthio)-1,3-dithiolane-2-thione

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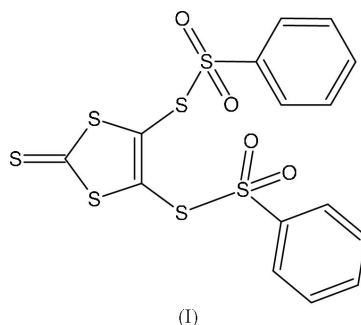
## Key indicators

Single-crystal X-ray study  
*T* = 293 K  
Mean  $\sigma(\text{C}-\text{C}) = 0.014 \text{ \AA}$   
*R* factor = 0.066  
*wR* factor = 0.185  
Data-to-parameter ratio = 16.0For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

The central 1,3-dithiolane-2-thione fragment of the title compound,  $\text{C}_{15}\text{H}_{10}\text{O}_4\text{S}_7$ , is planar; the conformations of the phenylsulfonylthio substituents in positions 4 and 5 of the dithiolane ring are related by an approximate twofold axis coinciding with the  $\text{C}=\text{S}$  bond of the 1,3-dithiolane-2-thione group. The planes of the phenyl rings form dihedral angles of 25.4 (4) and 31.1 (4)° with the least-squares plane of the central dithiolane ring.

## Comment

TTF (tetrathiafulvalene) and BEDT-TTF [bis(ethylene-dithio)tetrathiafulvalene] derivatives and their charge-transfer salts have attracted considerable interest because of their high electronic conductivity or superconductivity (Williams *et al.*, 1992). Furthermore, these compounds have received some attention in the third-order nonlinear optical field as a result of their extensively conjugated structure (Huggard & Blau, 1987). In order to obtain materials with high optical and/or electrical properties, and to investigate structure–property relationships, compounds with various groups bonded to sulfur in the 4- and 5-positions of the dithiolane ring are being studied extensively. As a part of these studies, the title compound, (I), the precursor of new TTF and BEDT-TTF derivatives, was prepared.



The molecular structure of (I), along with the atom-numbering scheme, is shown in Fig. 1. Thione atom S1 is coplanar with the dithiolane ring, and the  $\text{C1}=\text{S1}$  distance [1.622 (7) Å] is almost identical to that of a typical  $\text{C}=\text{S}$  double bond (1.60 Å; Allen *et al.*, 1987). The  $\text{C}-\text{S}$  bond lengths in the five-membered ring (1.722–1.749 Å), on the other hand, are shorter than a typical  $\text{C}-\text{S}$  single bond (1.82 Å; Allen *et al.*, 1987). These values show the high degree of conjugation in the five-membered ring of the title compound.

The 1,3-dithiolane-2-thione fragment is planar; the conformations of the phenylsulfonylthio substituents are related by

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an approximate twofold axis coinciding with the C1=S1 bond. The planes of the C4–C9 and C10–C15 phenyl rings form dihedral angles of 25.4 (4) and 31.1 (4)°, respectively, with the least-squares plane of the central dithiolane ring.

The crystal packing of the title compound does not show any specific interactions. A packing diagram is shown in Fig. 2.

## Experimental

The title compound was prepared by the reaction of  $(\text{Bu}_4\text{N})_2[\text{Zn}(\text{dmit})_2]$  (dmit is the 2-thioxo-1,3-dithiolane-4,5-dithiolate dianion,  $\text{C}_3\text{S}_3^{2-}$ ) and benzenesulfonyl chloride in acetone. Benzenesulfonyl chloride (60 mmol) was added dropwise to a solution of  $(\text{Bu}_4\text{N})_2[\text{Zn}(\text{dmit})_2]$  (15 mmol) in acetone (100 ml). The mixture was stirred for 2 h at room temperature and then dissolved in acetone (100 ml). The resulting orange precipitate was filtered off; the orange filtrate was then left to stand for several days at approximately 277 K, and yellow crystals used for the X-ray structure determination were obtained.

### Crystal data

$\text{C}_{15}\text{H}_{10}\text{O}_4\text{S}_7$   
 $M_r = 478.65$   
 Monoclinic,  $P2_1/c$   
 $a = 12.3011$  (17) Å  
 $b = 12.3846$  (13) Å  
 $c = 14.187$  (2) Å  
 $\beta = 114.338$  (11)°  
 $V = 1969.2$  (5) Å<sup>3</sup>  
 $Z = 4$

$D_x = 1.614$  Mg m<sup>-3</sup>  
 Mo  $K\alpha$  radiation  
 Cell parameters from 61 reflections  
 $\theta = 5.0$ – $12.4^\circ$   
 $\mu = 0.82$  mm<sup>-1</sup>  
 $T = 293$  (1) K  
 Prism, yellow  
 $0.35 \times 0.31 \times 0.22$  mm

### Data collection

Bruker P4 diffractometer  
 $\omega$  scans  
 Absorption correction:  $\psi$  scan  
 (XSCANS; Bruker, 1996)  
 $T_{\min} = 0.638$ ,  $T_{\max} = 0.835$   
 4637 measured reflections  
 3768 independent reflections  
 1760 reflections with  $I > 2\sigma(I)$

$R_{\text{int}} = 0.054$   
 $\theta_{\text{max}} = 26.0^\circ$   
 $h = -15 \rightarrow 1$   
 $k = -15 \rightarrow 1$   
 $l = -16 \rightarrow 17$   
 3 standard reflections  
 every 97 reflections  
 intensity decay: none

### Refinement

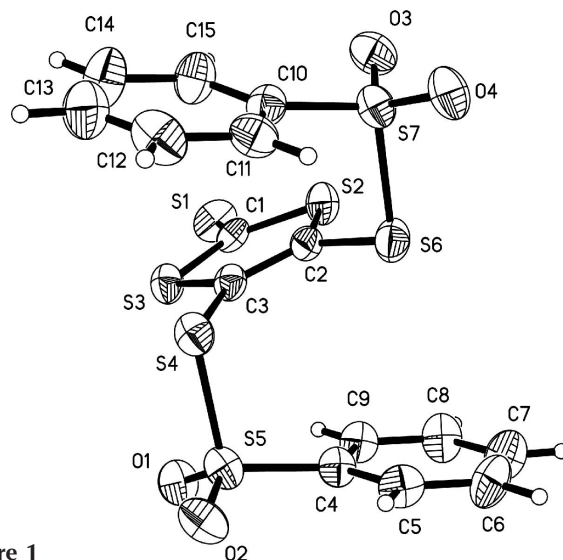
Refinement on  $F^2$   
 $R[F^2 > 2\sigma(F^2)] = 0.066$   
 $wR(F^2) = 0.185$   
 $S = 1.13$   
 3768 reflections  
 236 parameters  
 H-atom parameters constrained

$w = 1/[\sigma^2(F_o^2) + (0.02P)^2 + 8.6426P]$   
 where  $P = (F_o^2 + 2F_c^2)/3$   
 $(\Delta/\sigma)_{\text{max}} < 0.001$   
 $\Delta\rho_{\text{max}} = 0.45$  e Å<sup>-3</sup>  
 $\Delta\rho_{\text{min}} = -0.61$  e Å<sup>-3</sup>  
 Extinction correction: SHELXTL  
 Extinction coefficient: 0.0058 (5)

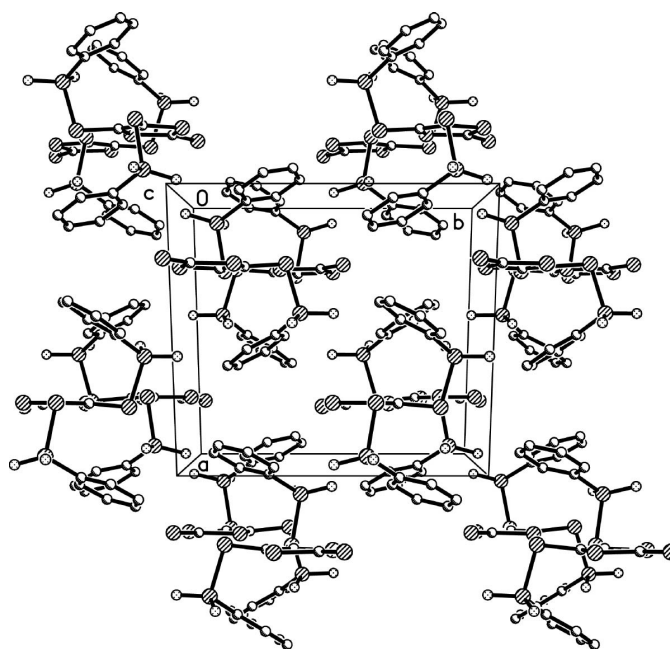
All H atoms, placed in geometrically calculated positions (C–H = 0.93 Å), were refined in the riding-model approximation; their  $U_{\text{iso}}(\text{H})$  values were set to 1.2 times  $U_{\text{eq}}$  of their parent atoms.

Data collection: XSCANS (Bruker, 1996); cell refinement: XSCANS; data reduction: SHELXTL (Bruker, 1997); program(s) used to solve structure: SIR97 (Altomare *et al.*, 1999); program(s) used to refine structure: SHELXTL; molecular graphics: SHELXTL; software used to prepare material for publication: SHELXTL.

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**Figure 1**  
 The molecular structure of the title compound, showing 50% probability displacement ellipsoids. H atoms are drawn as small spheres of arbitrary radii.



**Figure 2**  
 Packing diagram for the title compound viewed along the  $c$  axis. H atoms have been omitted.

## References

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